# Hydrodynamic and Water Quality Modeling for Chester River using SCHISM/HEM3D

by

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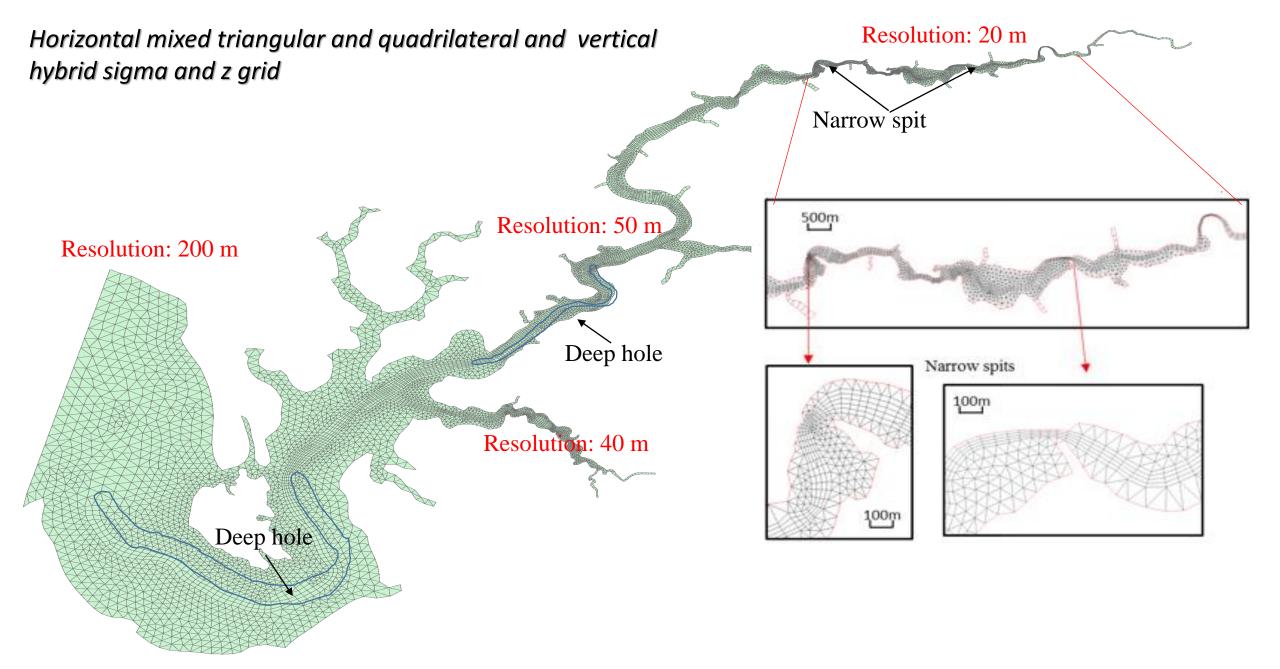


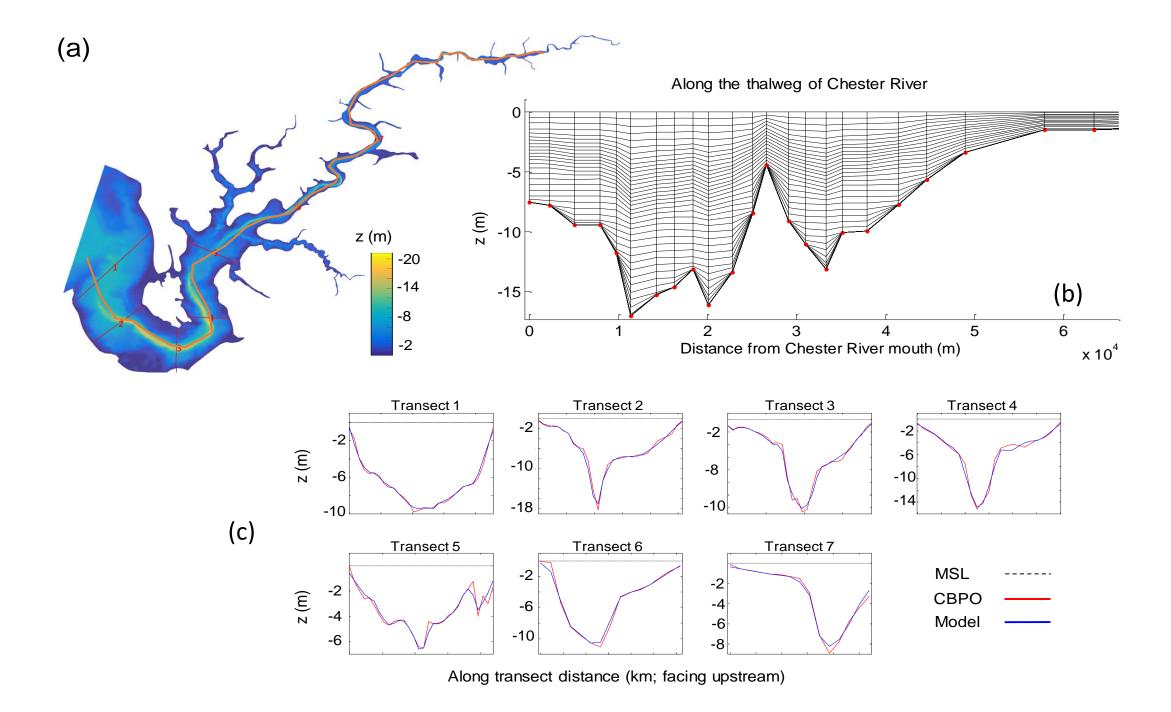
# Outline:

- Hydrodynamic modeling in the Chester River
  - 1) model domain and topography
  - 2) salt intrusion and temperature
  - 3) wind wave and suspended sediment
- II. Water quality processes -
  - 1) phytoplankton dynamics
  - 2) oxygen dynamics
- III. A PH dependent phosphorus release scheme
- IV. Conclusion

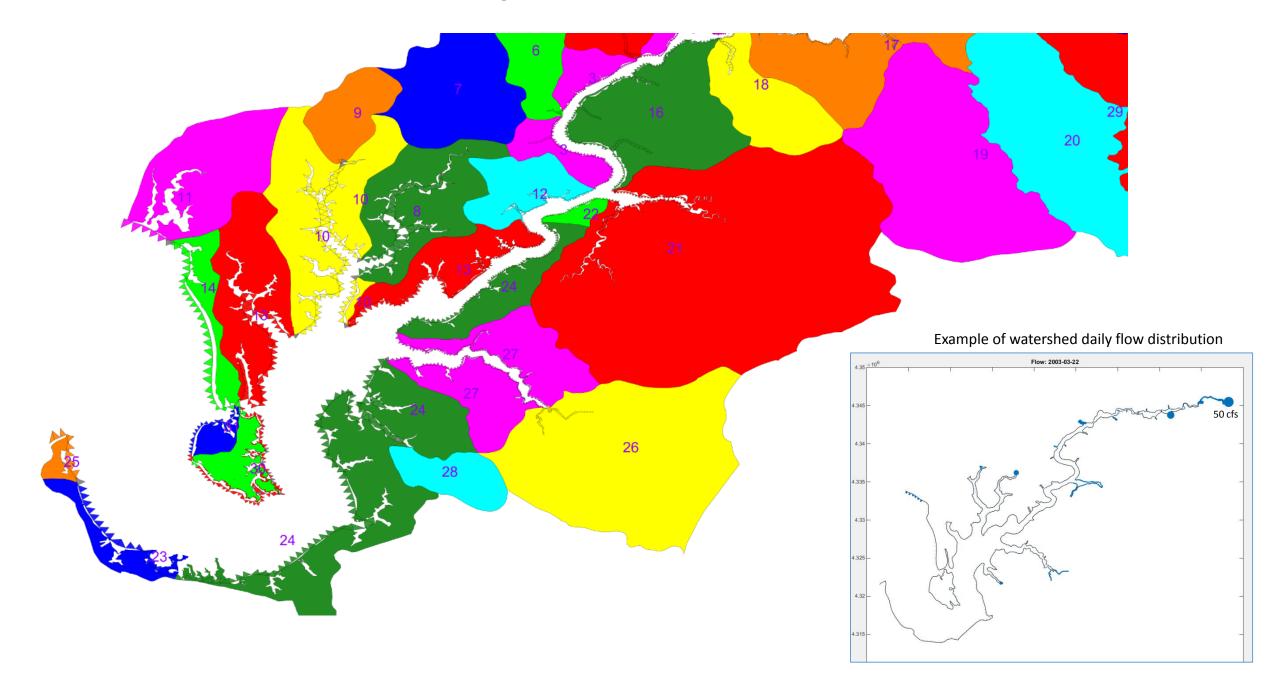


# I. Hydrodynamic modeling

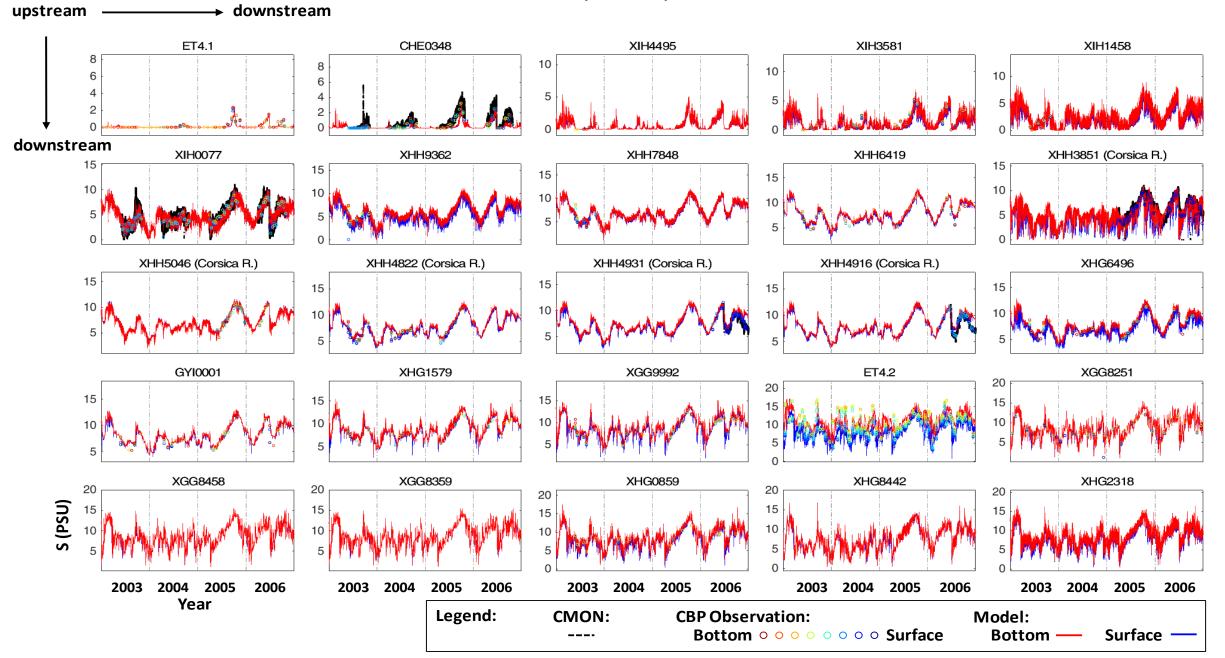




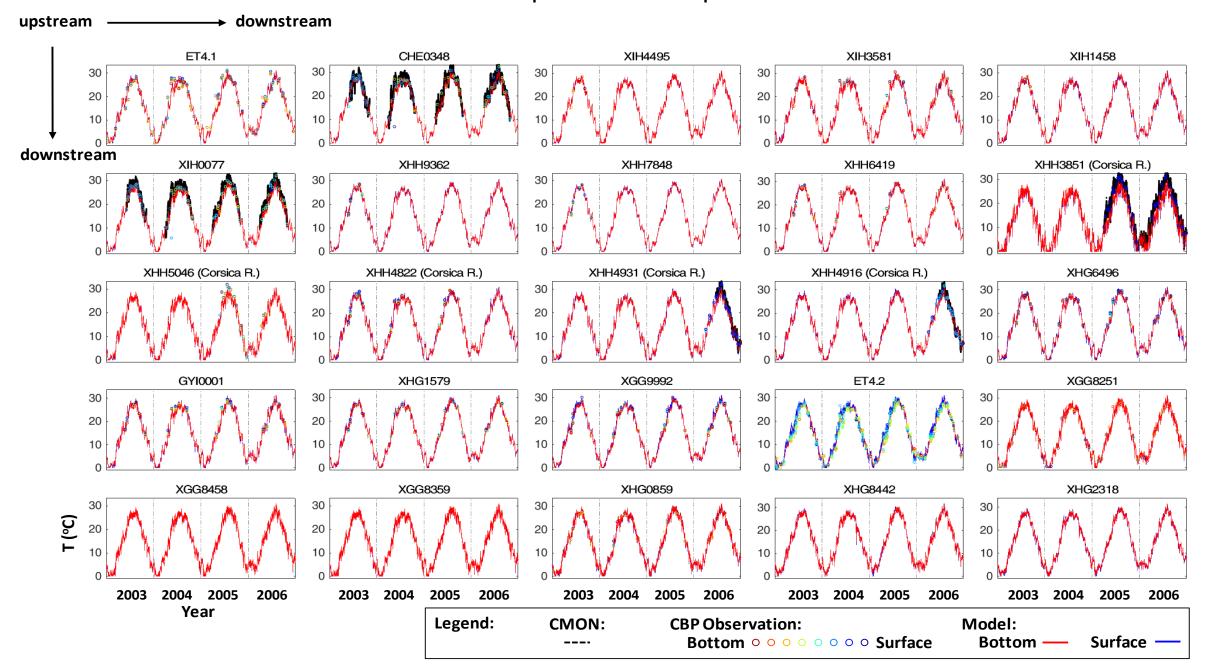
# Watershed delineation and loading



#### Salinity Comparison

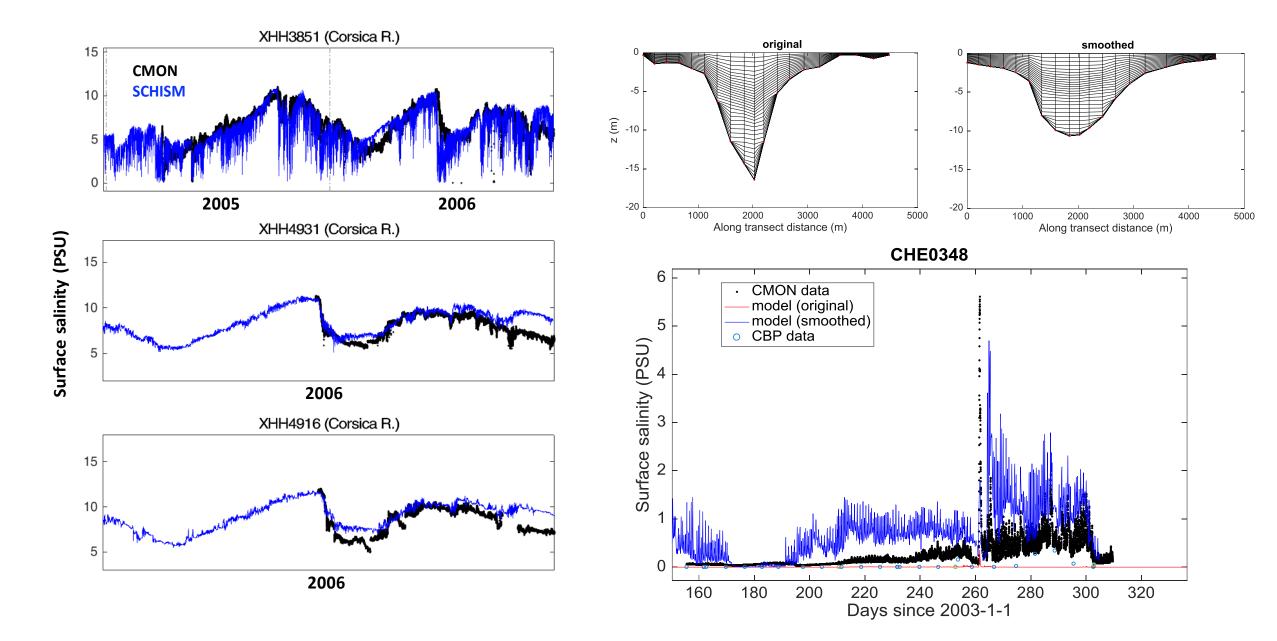


#### Temperature Comparison

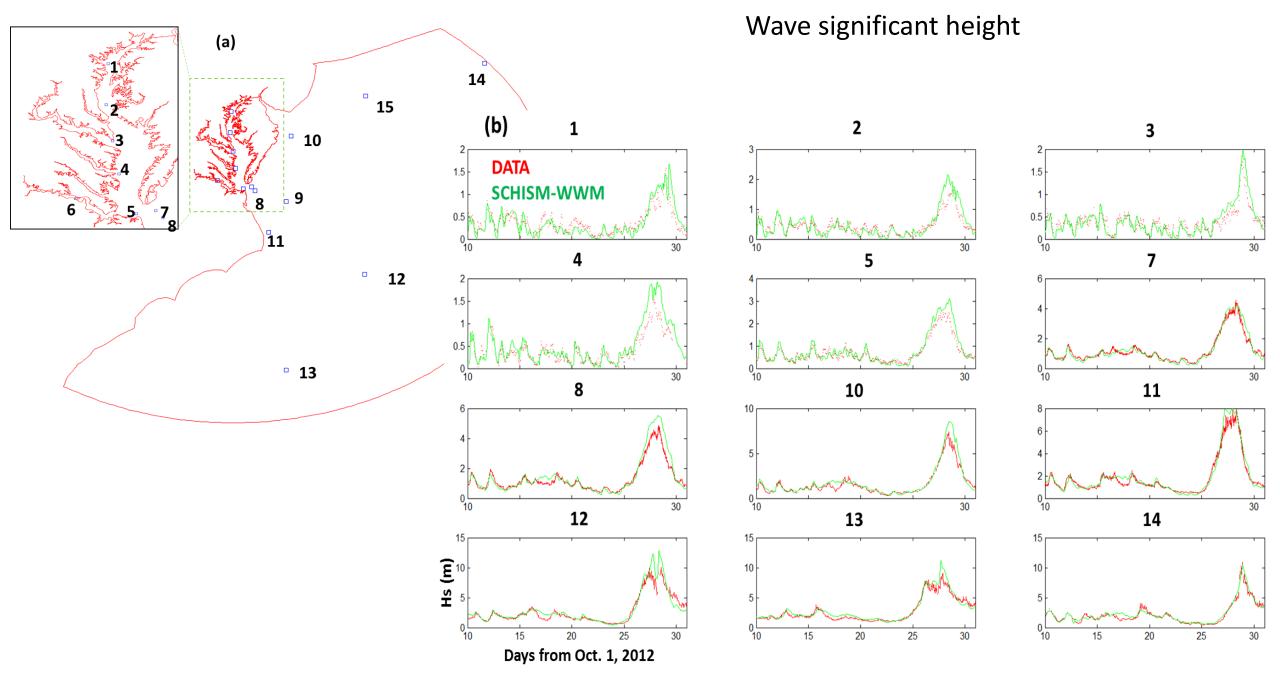


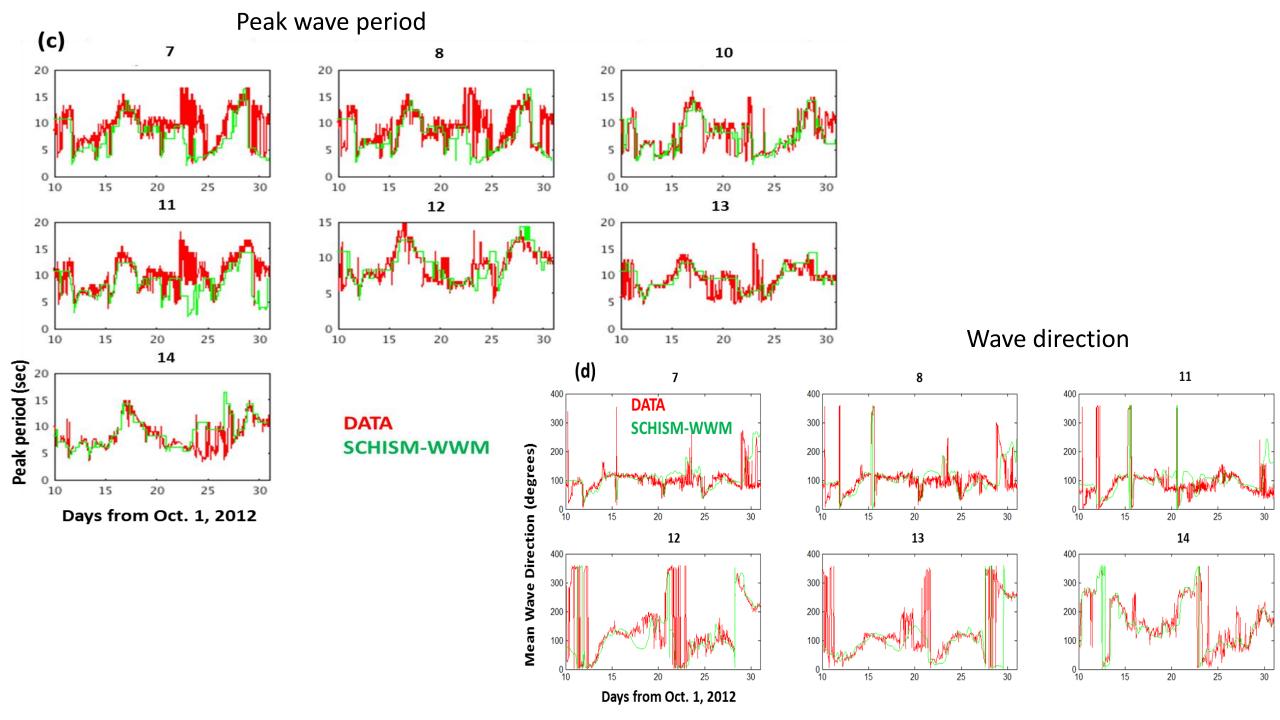
#### Comparison with Cmon salinity data

### Sensitivity to the cross-sectional discretization

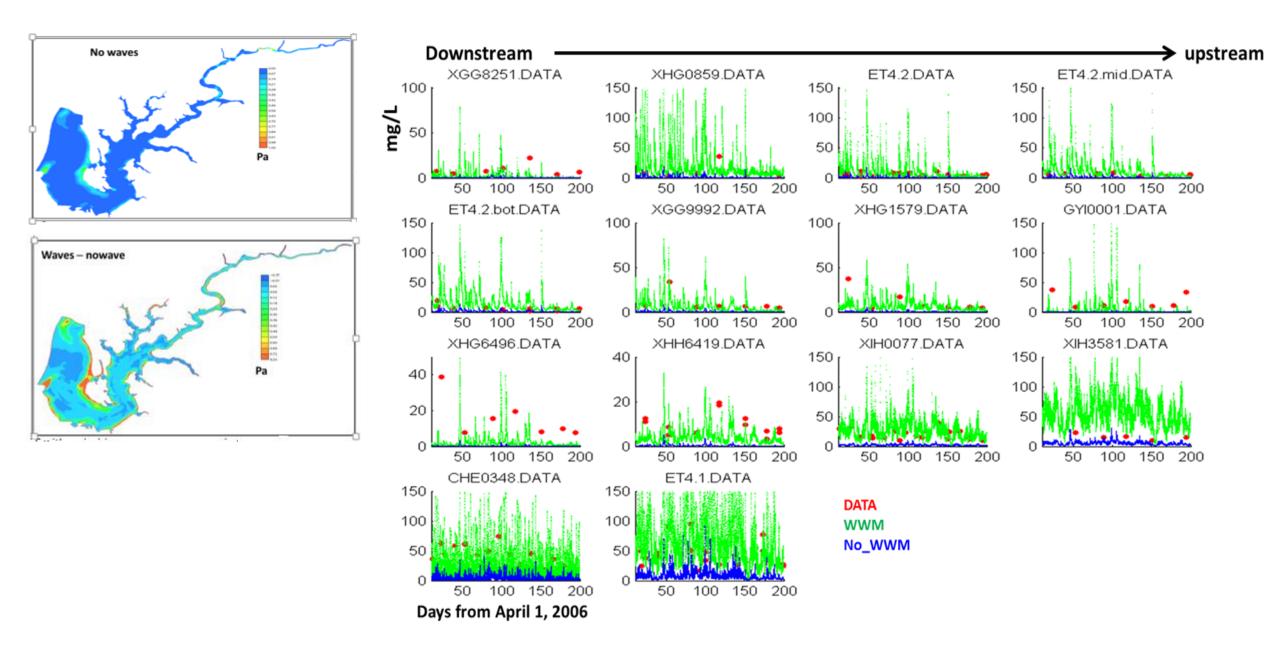


## Wind wave modeling

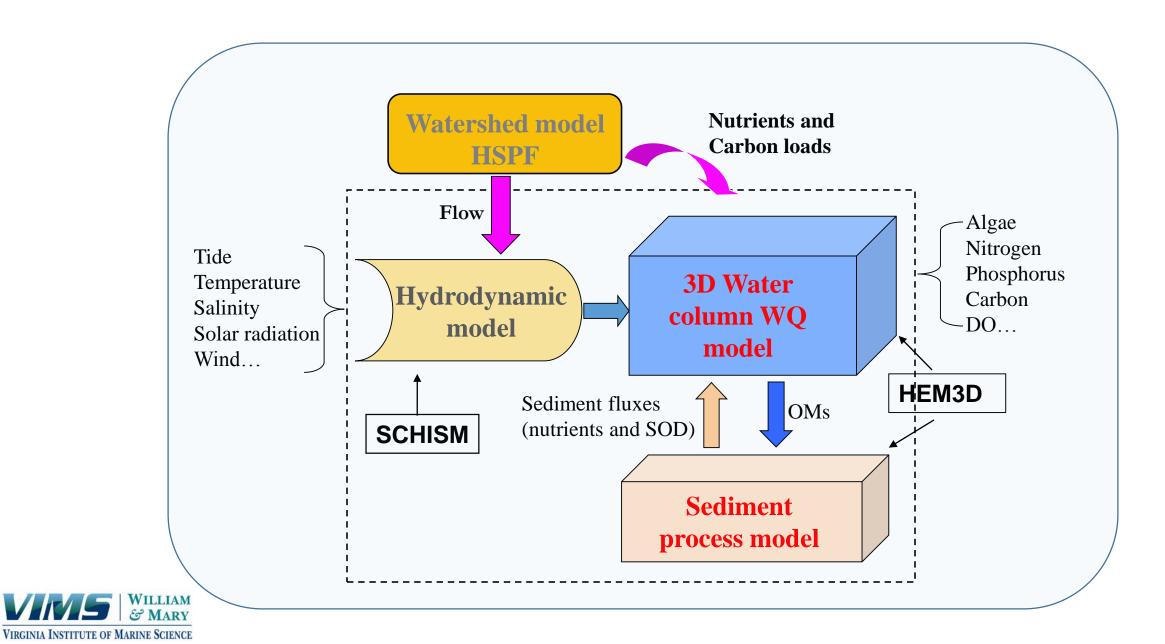




#### Inorganic component of TSS



# II. Water quality processes

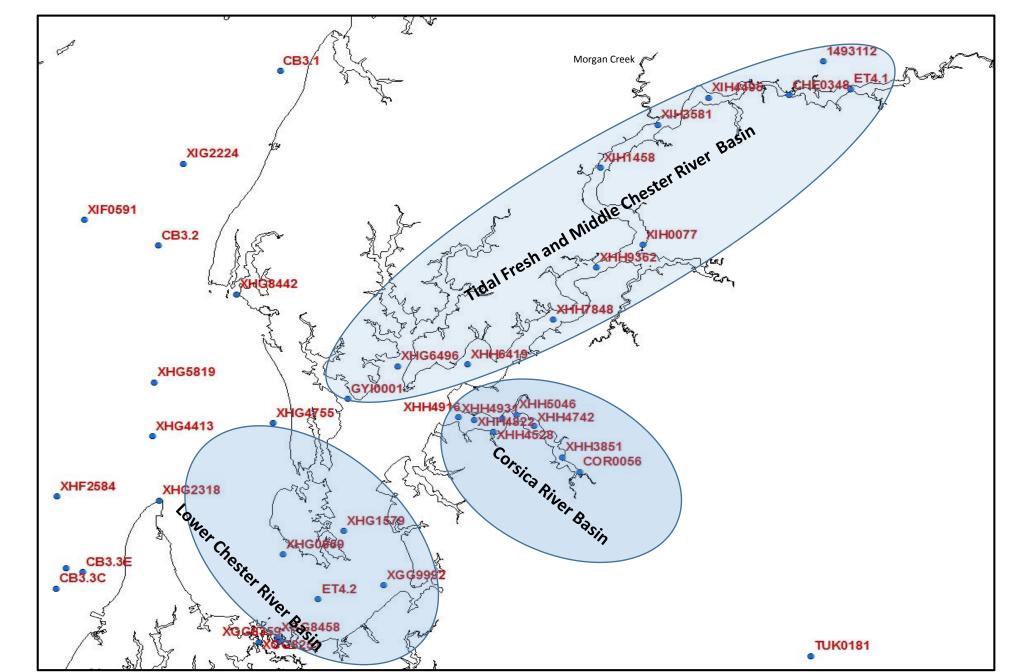


### Field Station Map

Divided into three major basins

- 1. Tidal fresh and middle Chester River basin
- 2. Corsica River basin
- 3. Lower Chester River basin

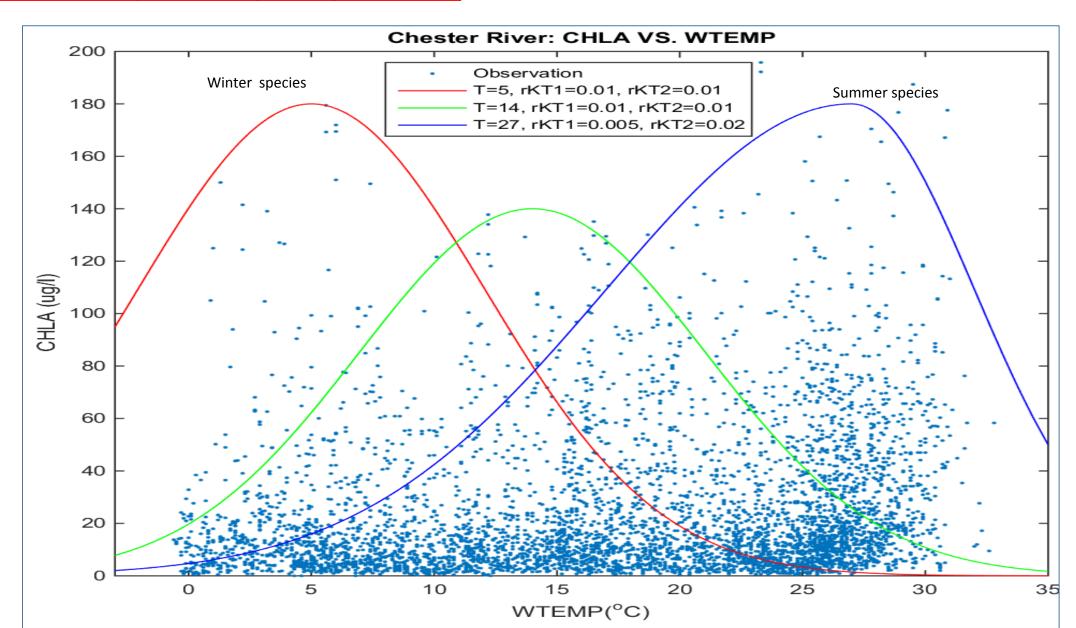
(used later for presenting modeling results)

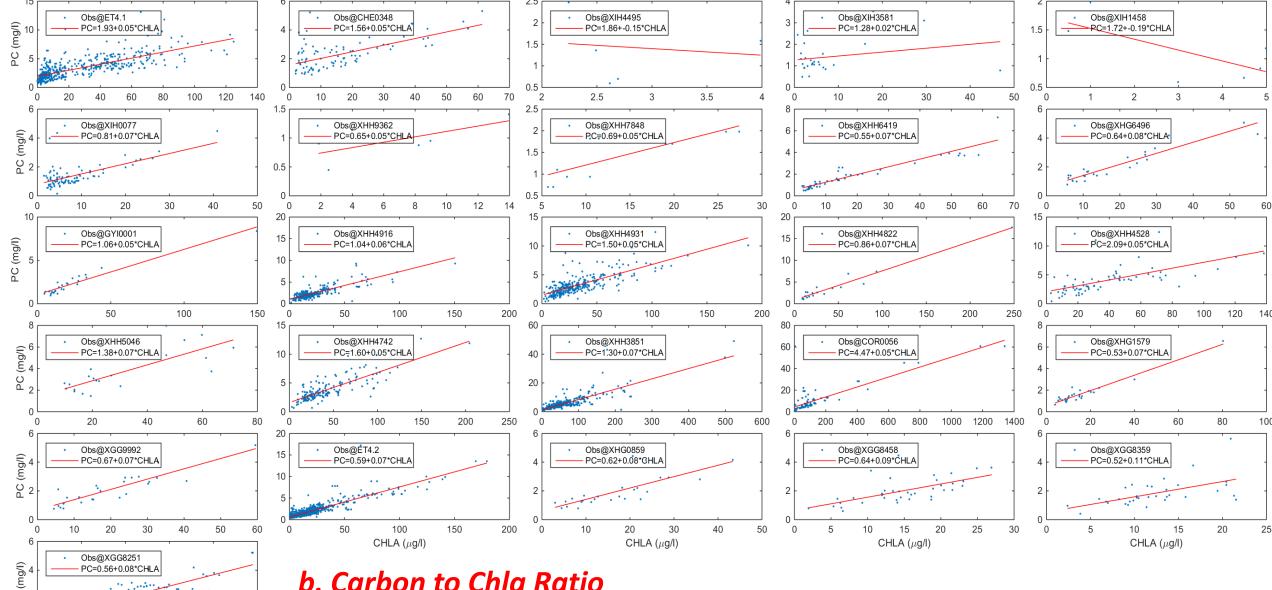




## Parameter specification

## a. Optimal temperature for algal bloom





### b. Carbon to Chla Ratio

CHLA (μg/l)

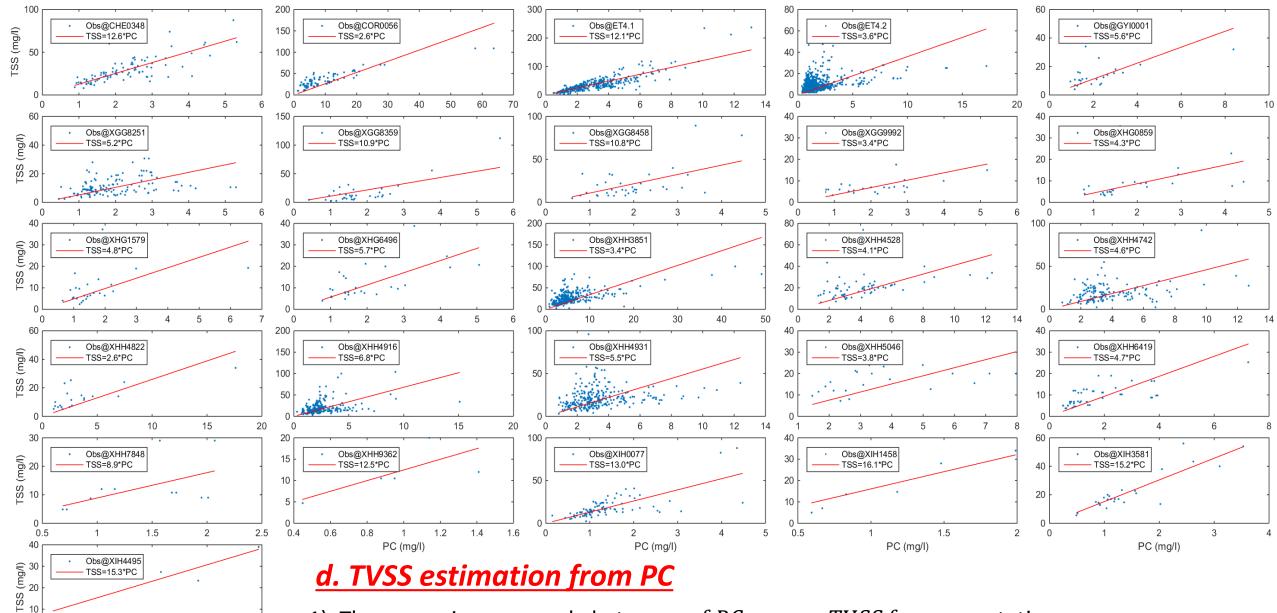
- 1). The analysis shows that the slope of Chla:C is about (50 70 in mg/l unit) in Chester River
- 2). A constant value of Chla to Carbon ratio =59 is used.

#### c. Light attenuation

The formulation for light attenuation is as follows:

$$Ke = a_0 + a_1 * CHLA + a_2 * TSS$$
 $Ke$ : light attenuation  $(m^{-1})$ 
 $a_0$ : background attenuation  $(m^{-1})$ 
 $a_1$ : attenuation coefficient for CHLA  $(\frac{m^2}{mg})$ 
 $a_2$ : attenuation coefficient for TSS  $(\frac{m^2}{g})$ 
where a0=0.26; a1=0.017; a2=0.07;

- 1) CHLA is calculated from the model.
- 2) TVSS is estimated based by PC (particulate carbon)



- 1). The regression was made between of *PC* versus *TVSS* for every stations
- 2). This relation is interpolated in the whole domain

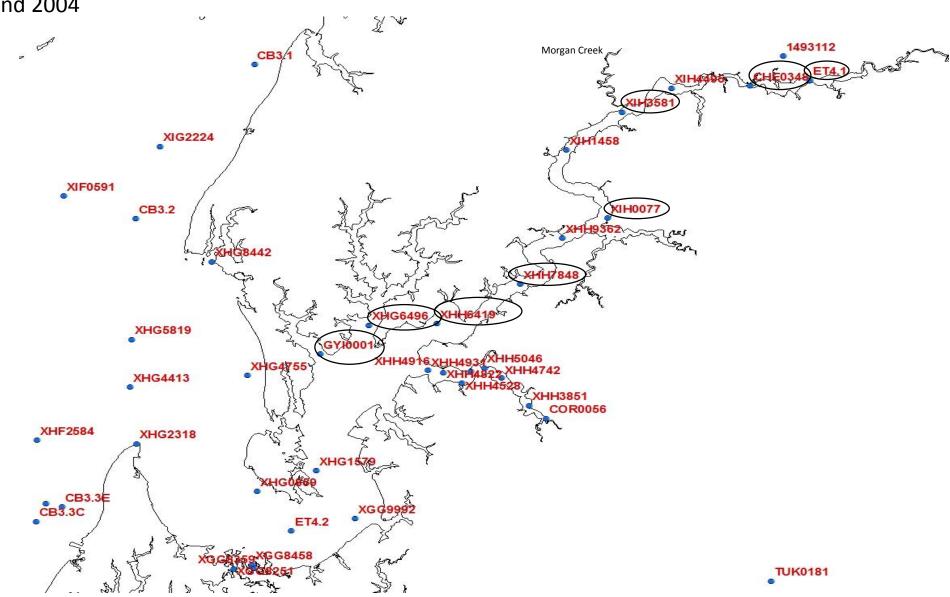
PC (mg/l)

3)\* Eventually, TSS is the addition of TVSS and Inorganic sediment from SCHISM

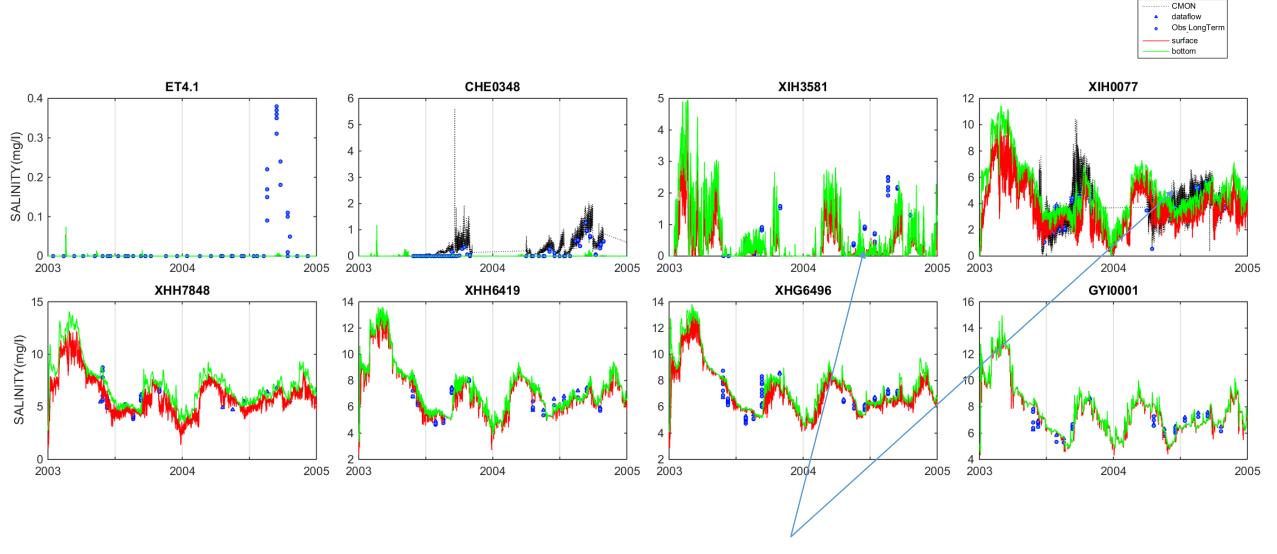
# Water quality modeling results

1. The water quality model is executed directly coupling with SCHISM. The time step used for both model is 120 sec.

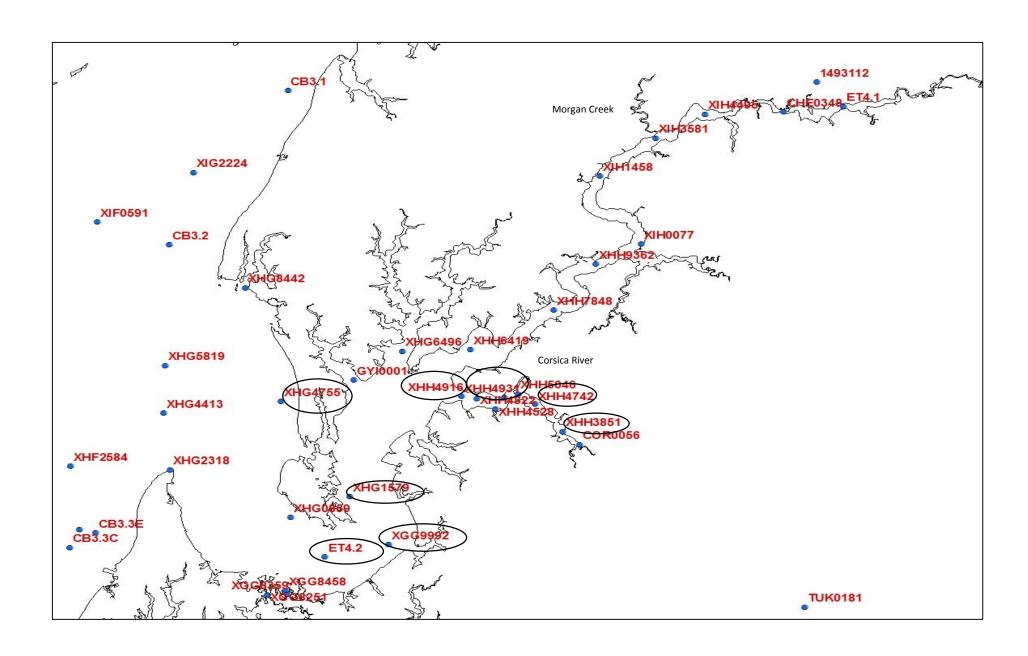
2. Simulating period 2003 and 2004



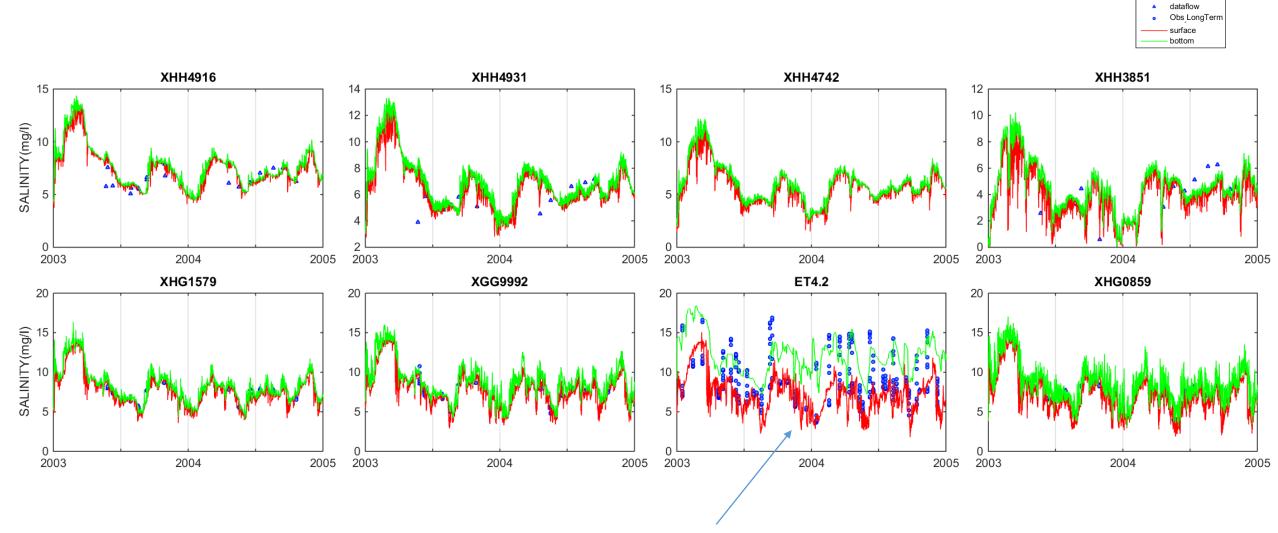
#### Salinity in the tidal fresh and middle Chester River



Salt intrusion prediction was better using coupled upper Bay open boundary condition

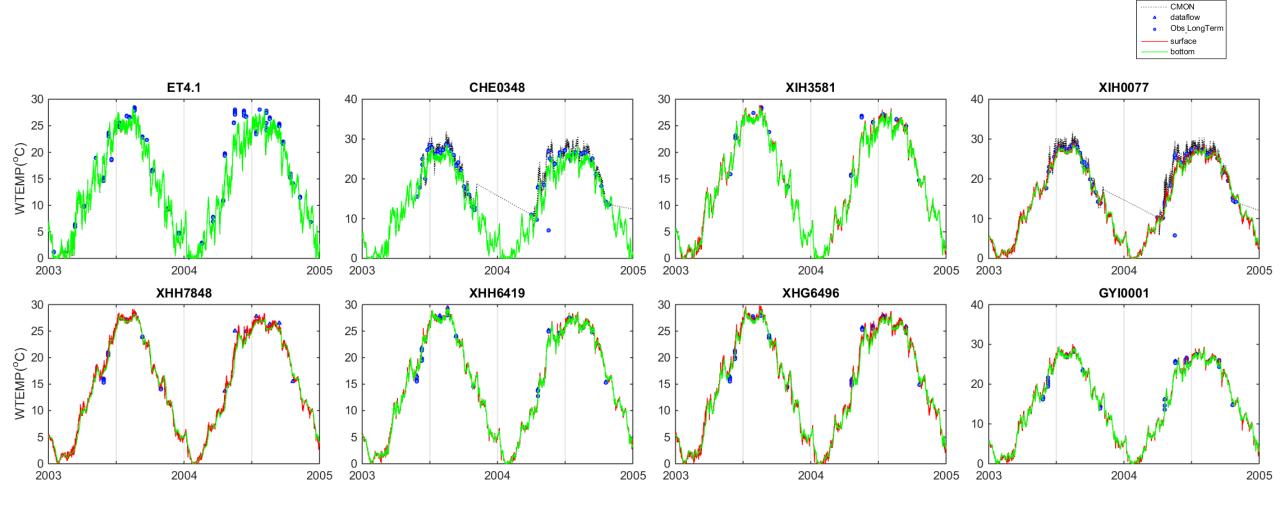


#### Salinity in the Corsica River and lower Chester River

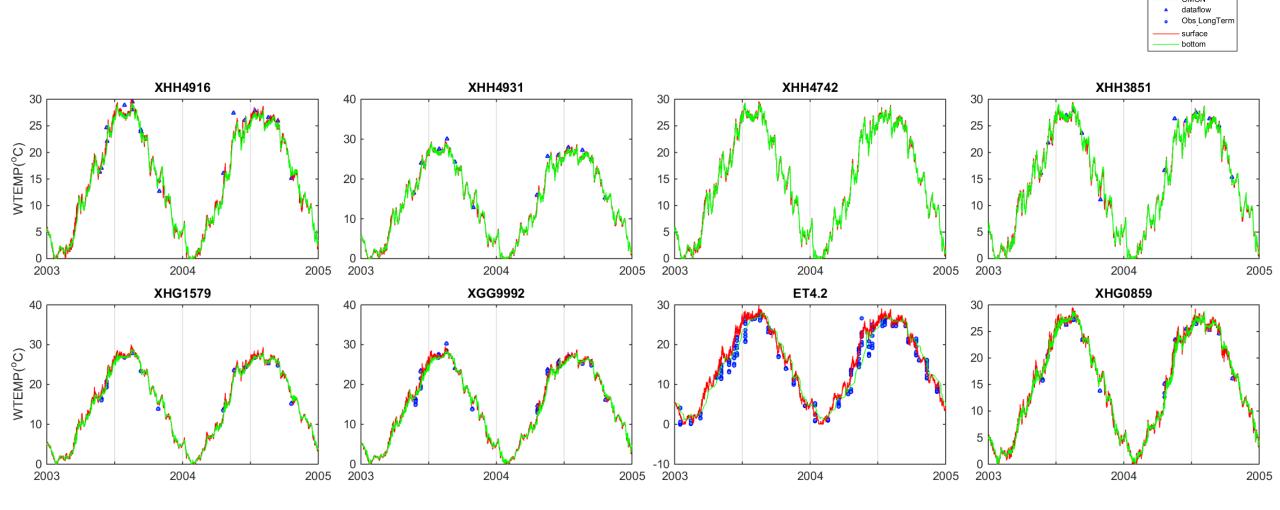


Salt intrusion prediction was better using coupled upper Bay open boundary condition

#### Temperature in the tidal fresh and middle Chester River

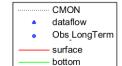


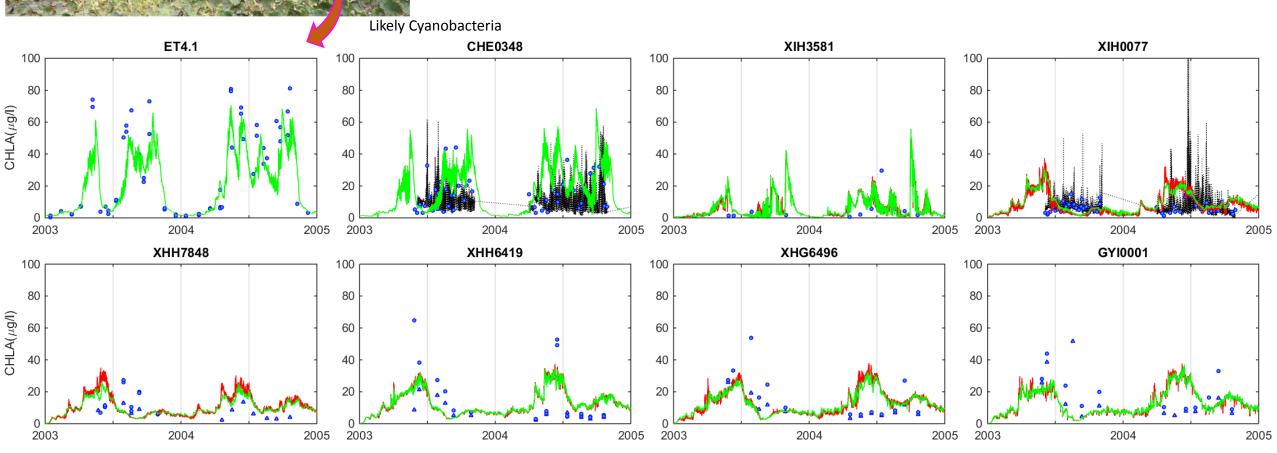
#### Temperature in the Corsica River and lower Chester River



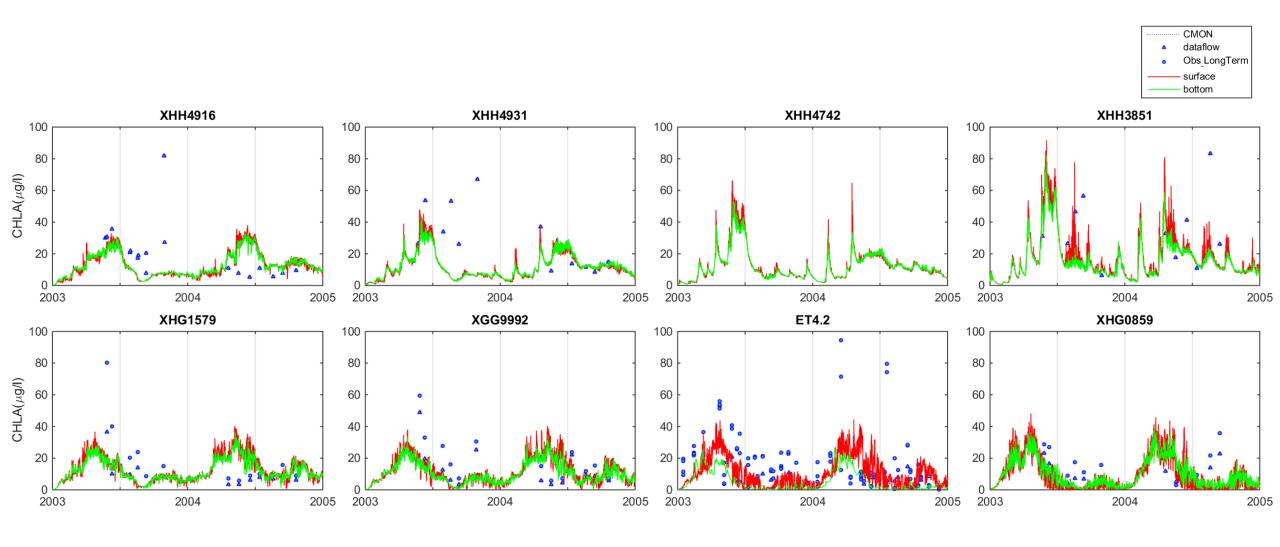
## Chlorophyll in the tidal fresh and middle Chester River



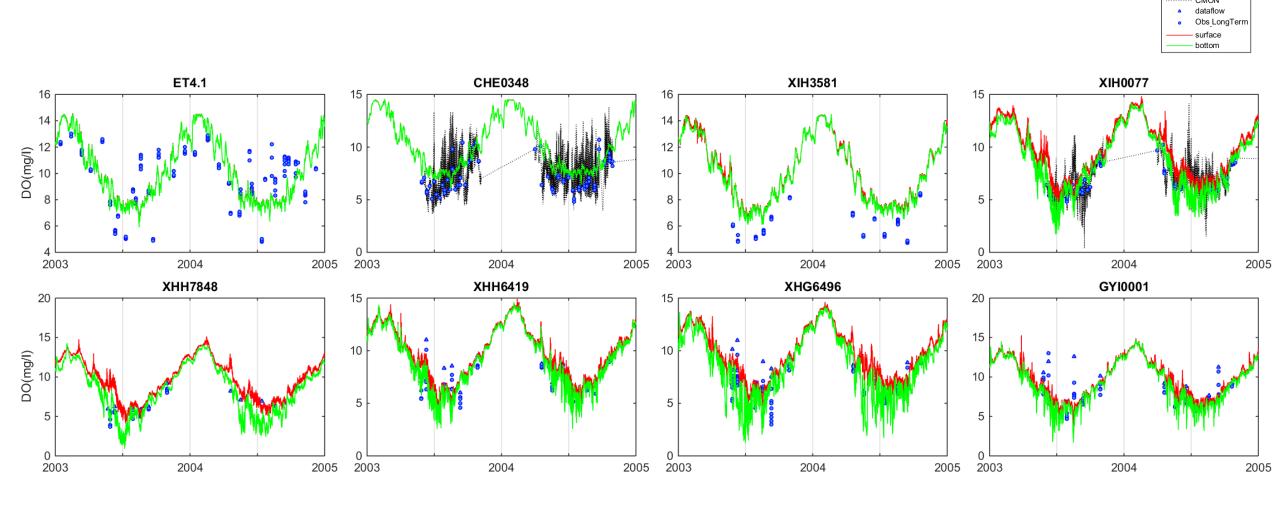




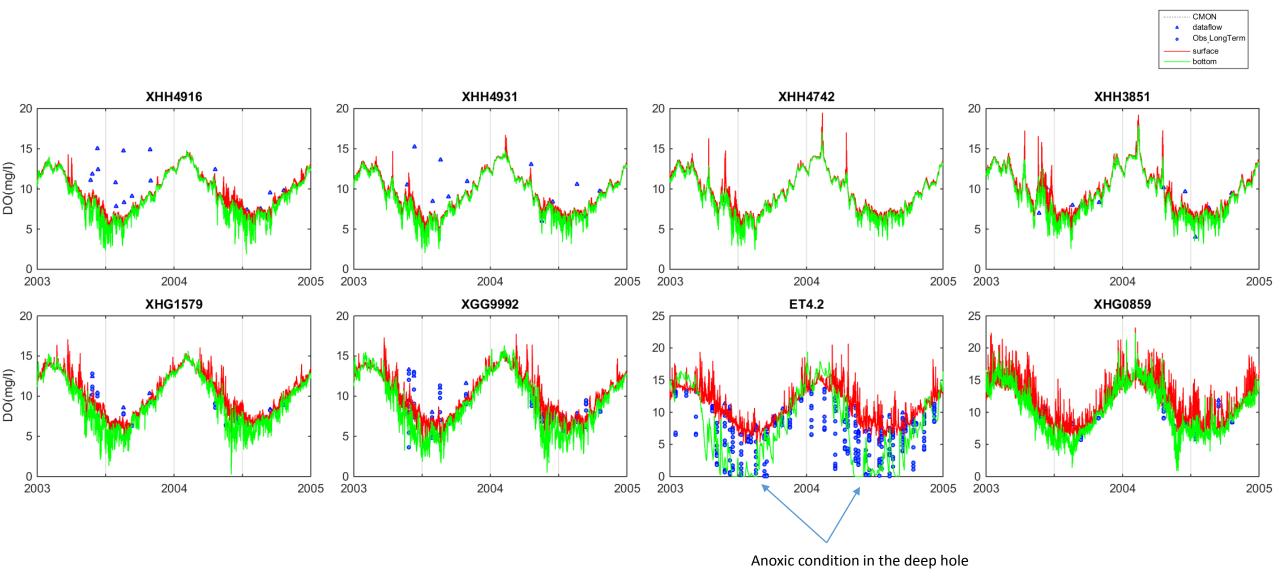
### Chlorophyll in the Corsica River and lower Chester River



### Dissolve oxygen in the tidal fresh and middle Chester River

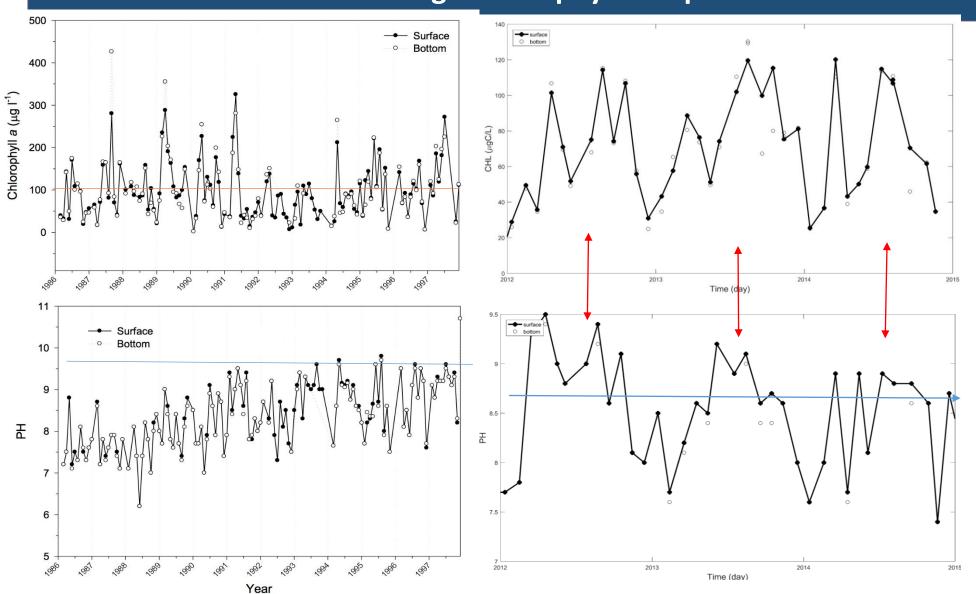


### Dissolve oxygen in the Corsica River and lower Chester River

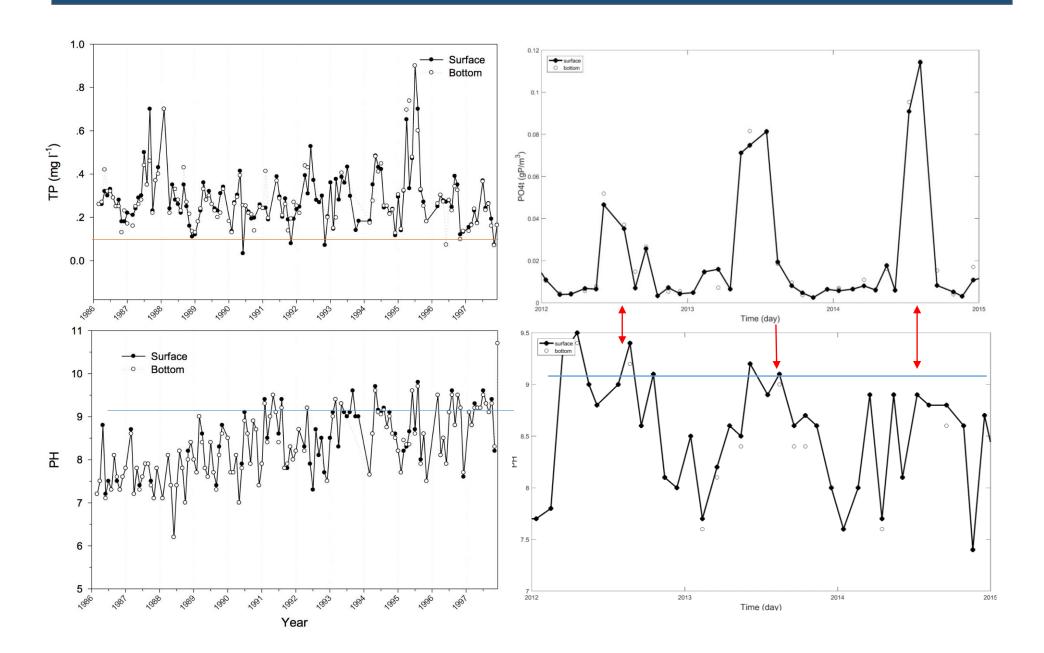


# III. A PH dependent phosphorus release scheme

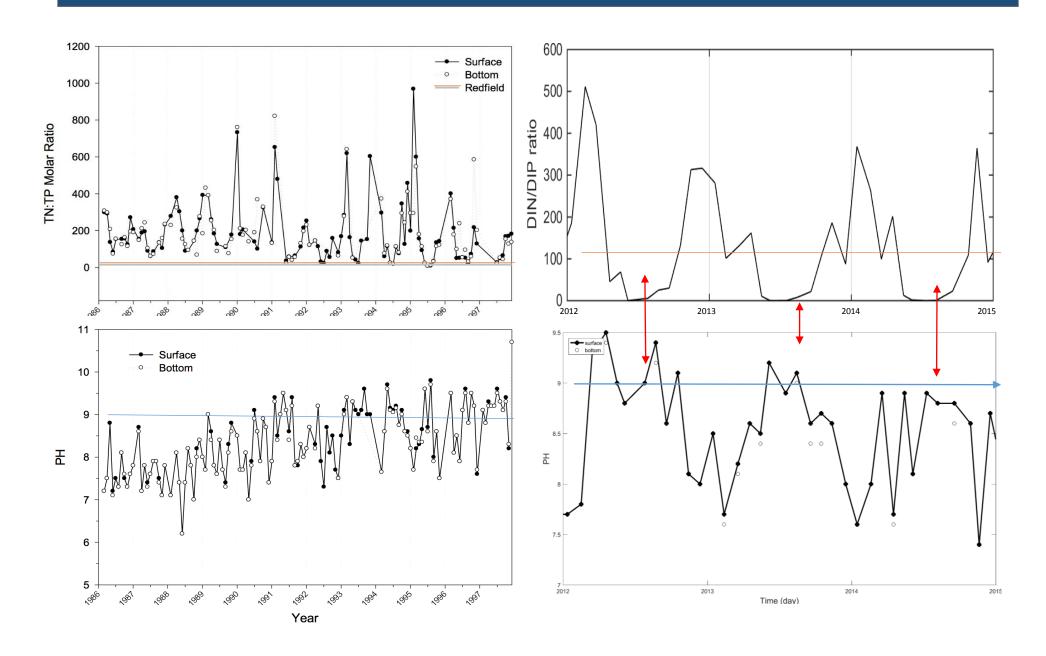
## Observational evidence of high chlorophyll and pH in Back River



## **Evidence of high phosphorus and pH in Back River**



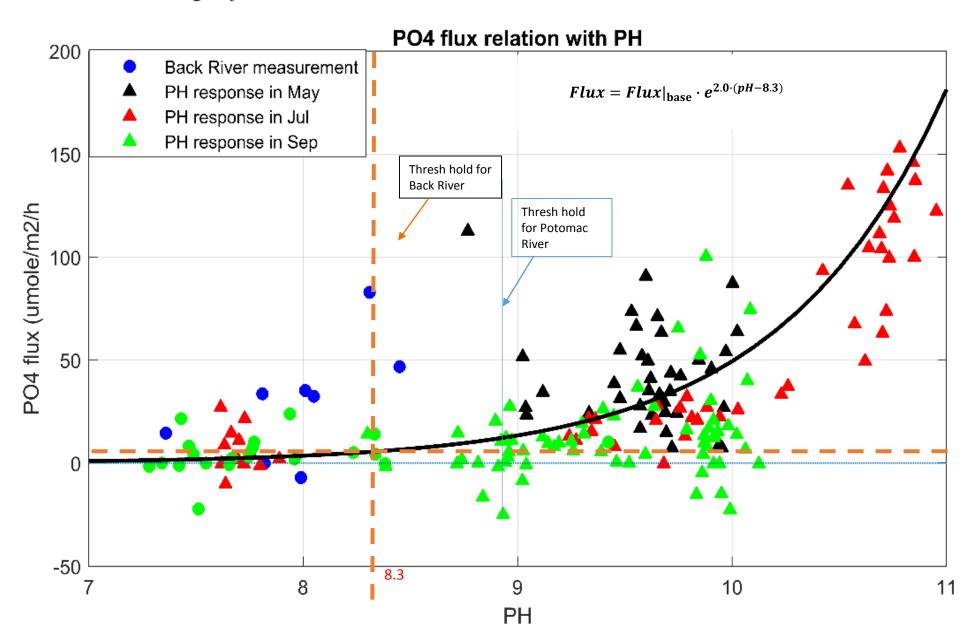
## Evidence of high N:P ratio versus low pH in Back River



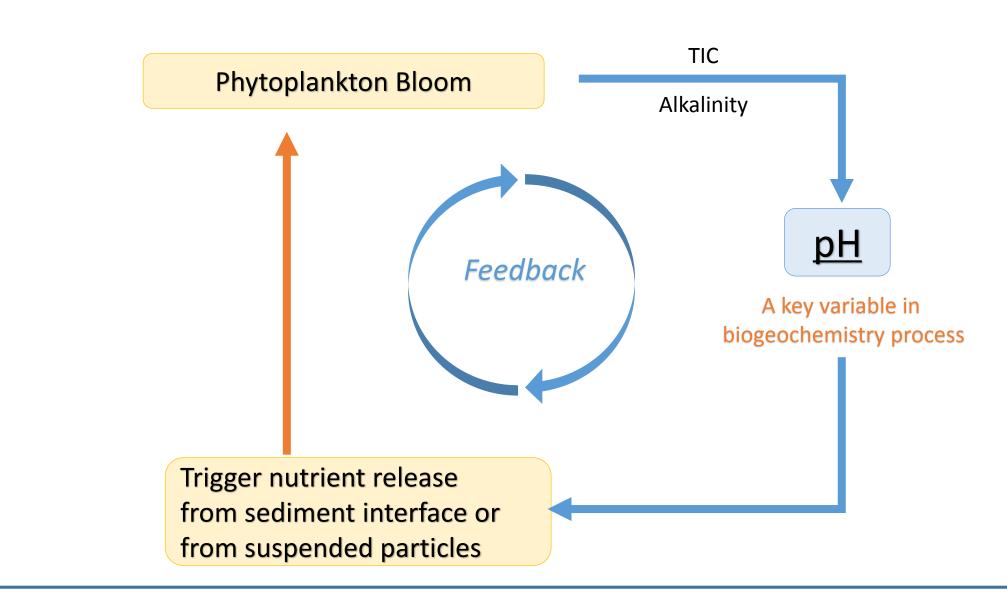
## Effect of pH on phosphorus release in Back River

- PH indeed poses effect on phosphorus

release in Back River with slightly lower thresh hold than that of Potomac River



The pH positive feedback loop is proposed as a mechanism for sustaining high algal biomass exceeding 100 -200  $\mu$ g/l in Back River



## **Development of pH sub-model**

Apply DIC and ALK from water quality model to the pH equation.

$$\frac{\partial \text{DIC}}{\partial t} = \sum_{x} (R - G) * PB_{x} + \frac{rKa * (CO2_{sat} - CO2)}{H} + rKHR * DOC$$

algal respiration and growth reaeration

DOC transformation

$$+\frac{SOD}{H*AOC} + rKCACO3*(CA_{sat} - CA)\frac{mC}{mCACO3}$$

sediment release caco3 formation/dissolution

$$\frac{\partial ALK}{\partial t} = \sum_{x} \left\{ -\frac{15}{14} * PN_{x} * G * PB_{x} + \frac{17}{16} * (1 - PN_{x}) * G * PB_{x} \right\}$$

algal uptake of NH4 algal uptake of NO3

$$*ANC_x*\frac{\text{mCACO3}}{2*\text{mN}} - rNit*NH4*\frac{\text{mCACO3}}{\text{mN}} + rKCACO3*(CA_{sat} - CA)$$

nitrification caco3 formation/dissolution

## **Development of pH sub-model**

#### Carbon equilibrium

$$H_2CO_3 \stackrel{K_1^*}{\rightleftharpoons} HCO_3^- + H^+ \cdots (1)$$

$$HCO_3^- \stackrel{K_2^*}{\rightleftharpoons} CO_3^{2-} + H^+ \cdots (2)$$

#### Mass balance of inorganic carbon

$$DIC = [CO_2] + [HCO_3^-] + [CO_3^{2-}] \cdots \cdots (3)$$

#### Water equilibrium

$$H_2O \stackrel{K_w^*}{\rightleftharpoons} H^+ + OH^-$$

#### Boron equilibrium

$$B(OH)_3 + H_2O \stackrel{K_B^*}{\rightleftharpoons} B(OH)_4^- + H^+$$

$$B_T = [B(OH)_3] + [B(OH)_4^-] = 4.16 \times 10^{-4} \times \frac{Salinity}{35}$$

#### **Approximation of alkalinity**

$$ALK \cong [HCO_3^-] + 2[CO_3^{2-}] + [B(OH)_4^-] + [OH^-] - [H^+] \cdots (4)$$

4 equations

6 variables

#### Governing equation: 5<sup>th</sup> order algebraic equation

$$DIC \cdot (K_B^* + h) \cdot (K_1h^2 + 2K_1K_2h) - [ALK(K_B^* + h)h - K_B^*B_Th - K_W^*(K_B^* + h) + (K_B^* + h)h^2] \cdot (h^2 + K_1h + K_1K_2) = 0$$

#### Numerically, solving the pH equation with Brent's Method

root bracketing and bisection



Stability

inverse quadratic interpolation.

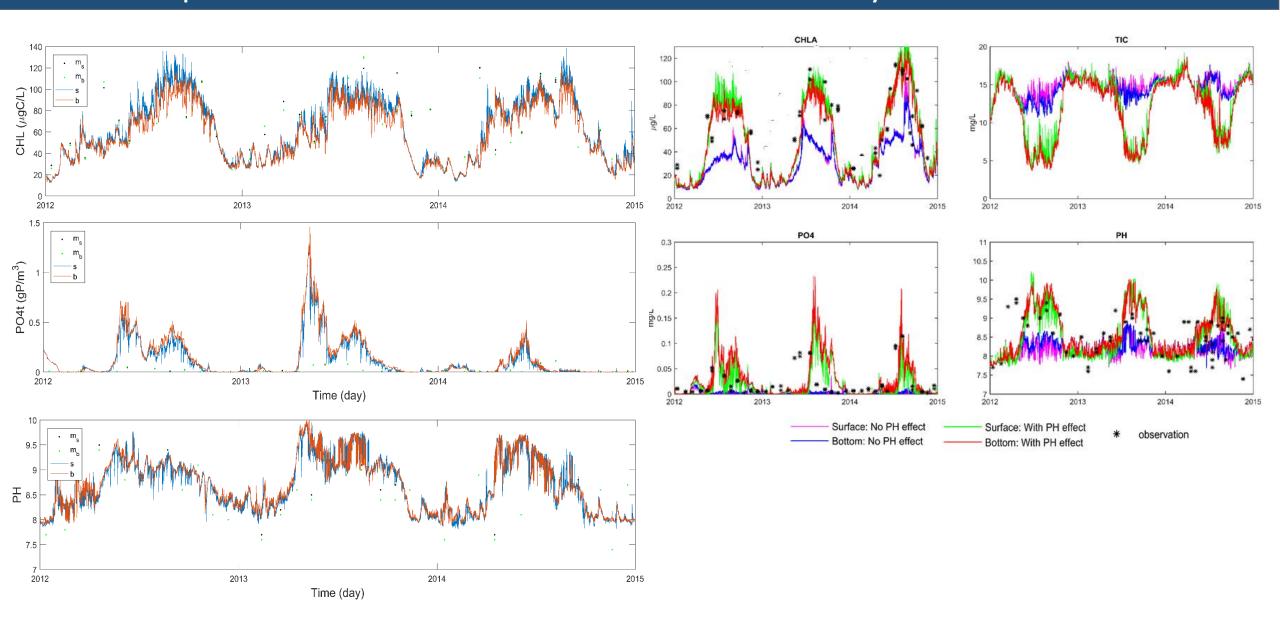


Efficiency (8.60298358  $\times$  10<sup>-7</sup>s per operation)

## **Results:**

Base comparison and calibration

#### • Sensitivity with and without PH



## IV. Conclusion

- The SCHISM hydrodynamic model goes a greater extent in treating the horizontal and vertical grids to ensure they reflect the real geometry truthfully.
- The modeled results of salinity and temperature compared very well with the observation when the coupled upper Bay open boundary condition was used.
- The simulation for the wind wave and inorganic sediment concentration showed the nearshore resuspension is important and can have significant impacts on the shallow water clarity.
- The water quality simulation was carried out based on the parameters derived from the rich data available. The TSS data was treated separately for the TVSS and inorganic suspended sediment. The latter was then provided by SCHISM.

# Conclusion (con't)

- The results for the chlorophyll simulation indicated that there is a persistent higher concentration occurred in the fresh water portion of the Chester in the summer, which is likely the bloom due to Cyanobacteria. Its concentration, however, drops quickly when encountering with the brackish water downstream.
- The DO anoxic condition occurred only in the deep hole in the lower Chester.
- A PH dependent phosphorus release scheme was developed which could be important for the algal dynamics in the fresh portion of the estuary, such as Back River or Corsica River.

